

Experimental Study of Hydrodynamics in the Aquarium Using PIV Method

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Abstract: The purpose of this study is to determine the phase indicator functions (vacuum rate, velocity and bubble size) of the gas-liquid flow. The gas-liquid flows in these columns (aquarium) are intrinsically unstable and the dynamics of such flows influence the mixing and mass transfer performance. It is therefore important to characterize the dynamics of gas-liquid flow. Also, the complete knowledge of the global dynamics of the fluids of the bubble column is based on that of the bubble. The experimental analysis is carried out using a two-phase instrumentation consisting of an optical fiber bi-probe. The use of the experimental techniques has enabled a better understanding of the hydrodynamics of two-phase flow. In terms of results, intrusive techniques provide local measurements while non-intrusive techniques provide a distribution over a cross-section with different spatial and temporal resolutions. The optical fiber bi-probe placed between two column flanges permit to have a complete mapping of the dispersed phase flow. The use of a mass flow meter and an ultrasonic flow meter, in different flow configurations, made it possible to obtain data on the operation of the column. However, the analysis of granulometry of the bubbles in the columns is performed by intrusive, flow-disrupting and non-intrusive techniques. Knowledge of bubble size and vacuum rate is crucial for determining interfacial air.

Keywords: Flow, Hydrodynamics, PIV, Aquarium, Bubbles

1. Introduction

Flow visualization provides insight into the interpretation of experimental results with respect to local measurements (PIV). The advantages and disadvantages of the various intrusive and non-intrusive techniques and their applications to multiphase flows have been presented by several previous works [1-3]. Indeed, as we will detail in this study, Taylor's flow is sensitive to its history, which raises the question of the reproducibility of the experiments and the measurements obtained. Also, to answer this problem, a technique of visualization of the flow was used in our experiment to determine experimentally the fields of velocity, the diameters of the bubbles of CO₂ and the rate of vacuum following the vertical one in an aquarium allowing to select the desired

hydrodynamic regime [4-6].

2. Description of the Experience

Principle

The technique consists in seeding the flow with DANTEC DYNAMICS particles. These particles are spherical crystals HGS-10, hollow Ø10µm. They are used at a wavelength around 500 µm. It is a crystalline microscopic powder with a diameter of 50 µm which has the particularity of aligning itself along the current lines. In the presence of an incident light, they reflect the light with deferential intensities according to the exposed surface. This gives an idea of the general topology of the flow.

A camera placed perpendicularly to the measurement plane

records the snapshots of the plotters. The images are then divided into interrogation cells. A cell of an image is mapped to the same cell in the twin image. An intercorrelation is then applied between the two cells, and the position of its maximum gives the value of the most probable displacement of all the particles. This procedure, applied to all the interrogation cells, makes it possible to obtain the projection of a velocity vector field. The acquisition of numerous images will make it possible to calculate statistical magnitudes of the 1000 images for the speed in the studied plane. Which did not make it possible to do the image processing and also with the calculations of the averages and to remove it from the scrambles.

3. Materials and Protocol

3.1. Materials

The following materials are used to carry out the experiments at the fluid mechanics laboratory of INSA Lyon.

Aquarium size 25x25x70 cm;

Laser Neodymium-doped Yttrium Aluminum Garnet;

99.7% pure CO₂ bottle;
 PC for recording purpose;
 FastCam Photron SA4 camera;
 NAVITAR Zoom 6000 lens;
 Syringe Pump Lambda Vit-Fit;
 High precision scale balance (up to 200 g).

3.2. Protocol

1. Filling 15L of fresh water in the aquarium;
2. Positioning the camera facing the aquarium and connect it to the computer;
3. Lights illuminated on the aquarium;
4. Priming the CO₂ flow by the syringe pump, at a speed of 0.8 mm/min;
5. Adjusting the focal length of the lens relative to the laser plane;
6. Camera settings (acquisition rate and resolution);
7. Visualization of the bubbles on the screen, via the Photron software of the camera.

Figure 1 presents the experiment apparatus used for the experiment.

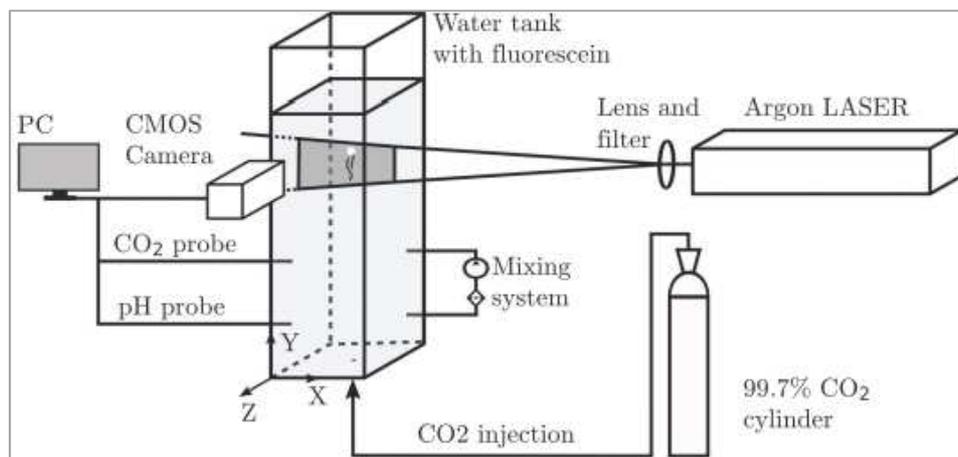


Figure 1. Experimental apparatus.

3.3. Fluorescein-Water Mixture

In the tank (figure 1) is poured a precise amount of water (demineralized water to eliminate any unknowns that could interfere in the study of the dissolution of CO₂), quantity measured using a volumetric flask of 2000 ml allowing to control at then the concentration of fluorescein to be poured. The fluorophore is then added to the water using a graduated pipette.

Once the substances are added, a B. Kern pump connected to the tank will suck and spit the mixture at two different places, allowing the liquid to be mixed cleanly so that it is homogeneous.

Light can be absorbed by chromophores and / or fluorophores. Fluorophore is a chromophore that emits photons when it absorbs light (figure 2). The absorption of a photon allows the passage of the chromophore from the fundamental energetic state to an excited state [7].



Figure 2. Visualization of the laser plane by the camera.

Beer-Lambert Law

When a beam of light passes through a solution of an

absorbing substance, it undergoes an absorption which is described by the Beer-Lambert law (Equation 1):

$$\frac{I(x)}{I(x=0)} = \exp\left(-2,303 \int_0^x \epsilon(\lambda)C(x')dx'\right) \quad (1)$$

Where I is the intensity of the light beam, x the distance along the light propagation path or optical path (x = 0 is an arbitrary choice), C the local concentration of the absorbing substance and Q its coefficient of absorption. The Beer-Lambert law is valid for a parallel and monochromatic beam of light of wavelength λ. The molecular absorption coefficient Q of the absorbing substance is a molecular property which reflects its ability to absorb a photon at a given wavelength.

4. Results

After each recording, a pattern figure 3 is placed in the aquarium instead of the diffuser, and an image is recorded in order to later measure the bubbles observed. The resolution is 1024 x 1024 for all shots. The chosen acquisition frequencies vary between 1000 img/s and 3000 img/s for the observation of moving bubbles, and is 50 img/s for the test pattern. The figure 3 shows the example of moving bubbles pattern.

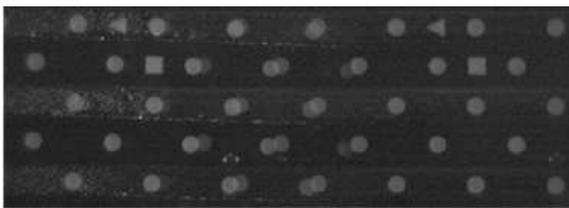


Figure 3. Example of the pattern.

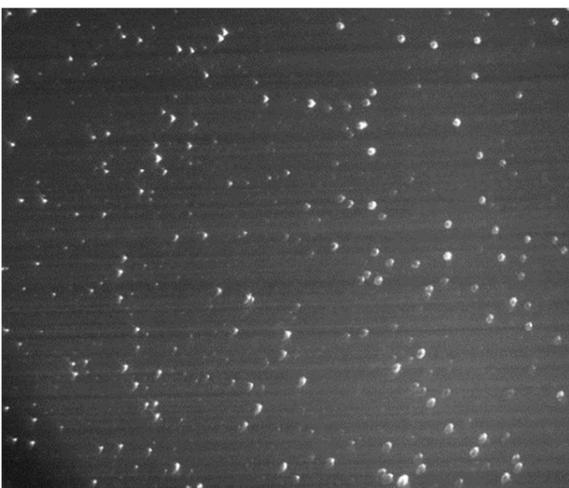


Figure 4. Averaged image with subtraction of porridge.

Acquisition of images

In order to obtain the images of the particles, the lighting of the flow is done using a laser sheet. This is produced by a lens system placed at the output of a Neodymium-doped Yttrium Aluminum Garnet laser source, emitting at a wavelength of λ = 532 nm. The pulse energy of the laser used

for PIV measurements is of the order of 25 mJ. The interest of the laser pulses is to avoid the drag effect related to the displacement of the particles during the exposure time of the camera. The images obtained enable to very clearly observe the crystals figure 4. The fast camera and its objective are much more powerful and high-performance to observe such small crystals.

4.1. Vacuum Rate

The analysis of figure 5 shows that the void fraction is directly proportional to the flow rate of CO₂ injected into the aquarium. For each CO₂ flow, the void ratio is relatively radially.

Gas retention is evaluated by static pressure measurements at different levels in the aquarium. These pressures are evaluated by a differential pressure probe (± 1 mbar) (Figure 1). The gas retention is then calculated by the method of [8] given by the equation. (2):

$$\epsilon_G = 1 - \frac{\Delta P}{\rho_L gh} \quad (2)$$

considering a density of the equal mixture:

$$\rho_m = \rho_G \epsilon_G + \rho_L \epsilon_L = \rho_L (1 - \epsilon_G) \quad (3)$$

where ε_G is the gas retention in the aquarium, ε_L is the liquid retention in the aquarium, ΔP is the pressure decrease between two measurements (Pa), h is the vertical height between two measurements (m), ρ_L is the density of the liquid phase (kg.m⁻³), ρ_G that of the gas phase (kg.m⁻³), ρ_m is the density of the mixture (kg.m⁻³) and g is the acceleration of gravity (ms⁻²). At a given gas flow, the gas retention is a function inverse to the size of the bubbles and generally an increasing function of the air flow.

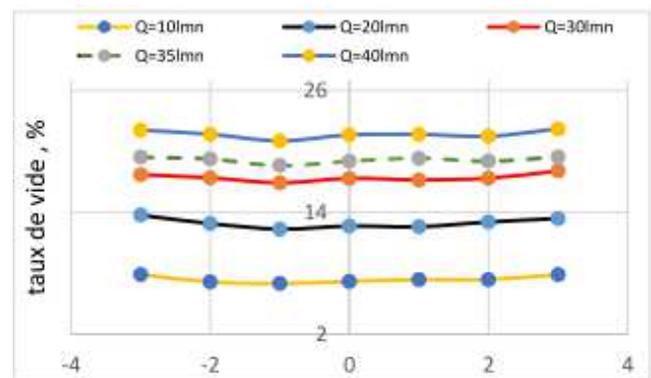


Figure 5. Evolution of the vacuum rate as a function of the injected CO₂ flow rate.

4.2. Bubble Size in Ascension

The aquarium is equipped with measuring sockets at different heights to introduce an optical bi-probe whose principle and use have been described by [9]. This probe is capable of measuring average bubble diameters, their rate of rise and the rate of gas retention. To facilitate the

interpretation of the data, certain hypotheses have been made: the movement of the bubbles is assumed exclusively vertical, the shape of the regular bubbles (spherical or ellipsoidal). However, with respect to the design of the probe, only bubbles larger than 2 mm could be taken into account in the evaluation of minimum diameter and average and maximum diameter (figure 6).

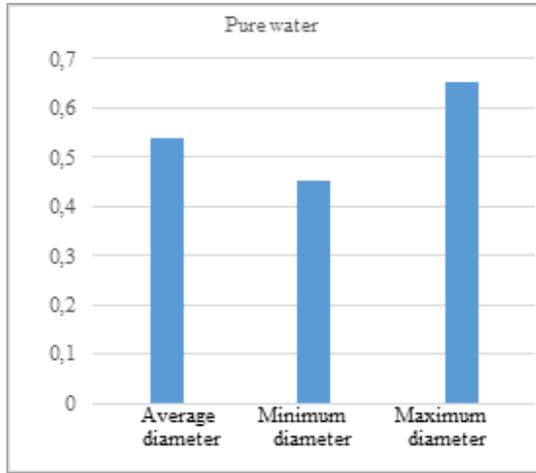


Figure 6. Size of the bubbles for the wooden diffuser.

4.3. Effect of CO₂ Flow on Bubble Size and Velocities

The mean diameters of the bubbles provided by the different types of freshwater injector in the aquarium are shown in Figure 7. They increase with the flow of CO₂ until reaching a limit value of about 0.7 mm [10]. In each case, the diameters of the bubbles are higher in fresh water. With a diffusion of the fine bubbles, the diameters measured by the probe are between 0.3 and 0.6 mm in fresh water, according to the air flow injected. With microbubble diffusion, the bubble diameters are between 0.4 and 0.6 mm in fresh water, and depending on the CO₂ flow. The variation of this speed with respect to the injected air flow is presented in figure 8. In fresh water, the speed varies linearly with HQ, between 0 and 0.7 m/s, whatever the type of injector used. The rise of microbubbles is slow compared to the fine bubbles in the aquarium.

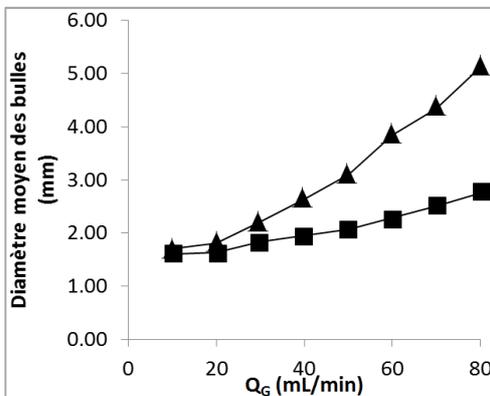


Figure 7. Average bubble diameter as a function of CO₂ flow.

By neglecting the increase in speed due to the

displacement of the water in the aquarium, the average speed of displacement of the bubbles can be calculated by the following relation:

$$U_G = Q_G / (\epsilon_G \cdot \Omega_i) \tag{4}$$

where U_G is the average bubble displacement velocity (m.s⁻¹), Q_G is the airflow (m³.s⁻¹), ε_G is the gas retention and Ω_i is the cross section of the inner pipe (m²).

The speed of movement of the bubbles is an increasing function of the air flow and the size of the bubbles.

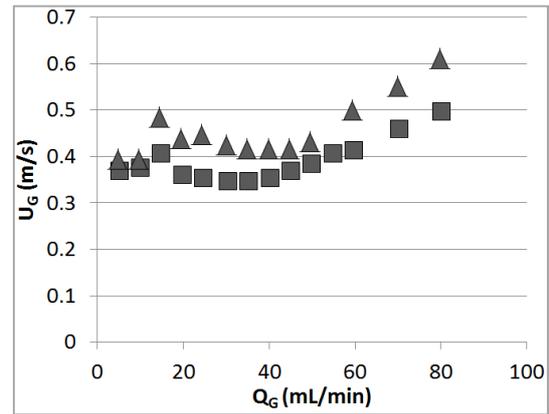


Figure 8. Bubble velocity versus CO₂ flow.

4.4. PIV Measures

In order to characterize the flow and the turbulence in the aquarium vertical velocity vectors are obtained using Velocimetry Particle Image (PIV) or Particle Image Velocimetry.

Frequency, exposure and recording range

The recording frequency of the DALSA camera is a function of the vertical recording height figure 9. However, the recording width does not influence the frequency. There must be a compromise between the recording range and the recording frequency. On the other hand if the recording frequency increases the maximum exposure time decreases. Exposure time greater than 2500µs results in blurred bubble outlines. On the other hand a significant exposure time allows a better brightness.

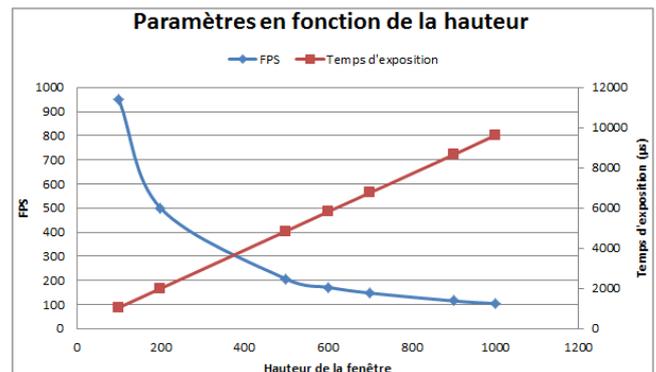


Figure 9. Recording Settings by Height.

These images are then processed to calculate the average

intensity, using a matlab program, giving the average gray level on a group of images. The results give the figure 10. On this curve, the peak of intensity appears very clearly for a concentration of $7.00E-06 \text{ mol}\cdot\text{L}^{-1}$, the intensity having gradually increased with the concentration until then. Past this peak, the intensity decreases quite linearly. This decay corresponds to fluorescein particles which hide themselves, blocking the light rays for high concentrations. Note a hook for a concentration of $1.30E-06 \text{ mol}\cdot\text{L}^{-1}$ fluorescein, which corresponds to a slight change on the diaphragm of the camera during handling.

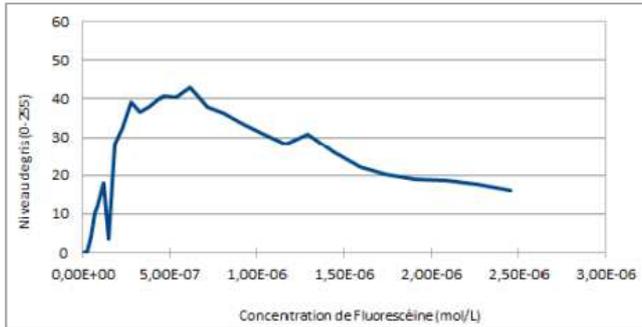


Figure 10. Intensity of fluorescein as a function of its concentration.

Fluorescein

Fluorescein is a dye which has the particularity of returning a fluorescent light when excited at a wavelength of 494 nm. However, this fluorescence depends on the dye concentration. One might think that the more concentrated it is, the more fluorescent it is. However, a study carried out under this project shows that after a stage, the fluorescence tends to disappear. The manipulation consists in gradually increasing the concentration of fluorescein in the tank, illuminated by a laser plane and recording the images. The images obtained have intensity varying with concentration (Figure 11).



a) with 120mL of fluorescein



b) with 490mL of fluorescein

Figure 11. Mean image with different amount of fluorescein.

Principle of the PIV

Particle Image Velocimetry is an optical method of determining the velocity of a fluid from the movement of particles carried by it. This technique allows to reconstruct a velocity field in a plane by dividing the images into interrogation windows and by determining the velocity of the particles located therein (figure 12). At present, optical techniques are preferred for turbulence measurements because the sensor is located outside the flow. They are said to be non-intrusive and do not disturb the flow [11].

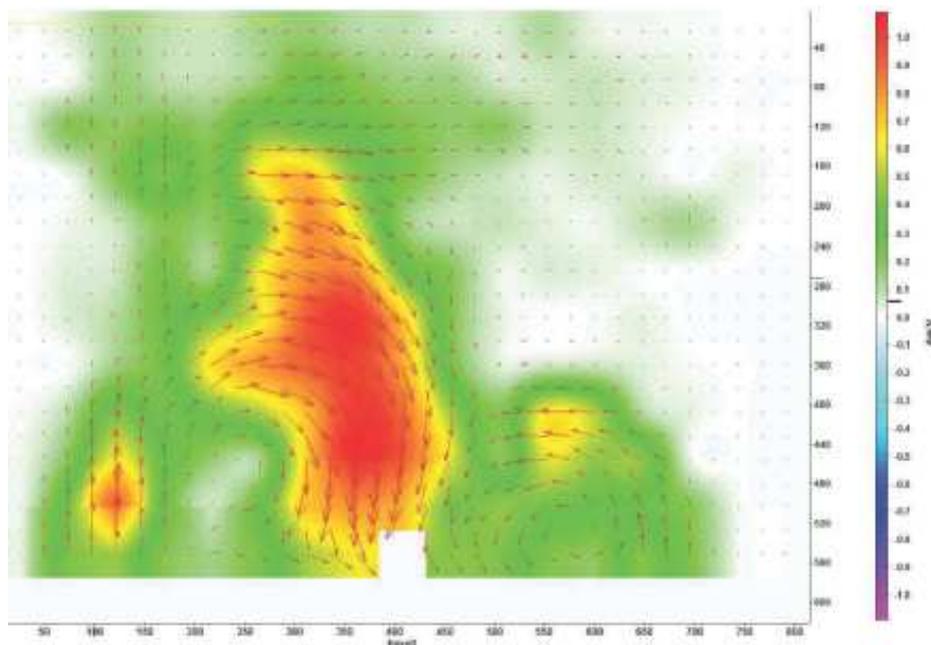


Figure 12. Velocity field measured by PIV.

5. Conclusion

The aquarium is a device that ensures a mixture between a bubble-shaped gas phase and a liquid phase. The use of PIV experimental techniques has allowed for a better understanding of the hydrodynamics of two-phase flow. For some decades, the studies carried out on this device have contributed to the understanding of their hydrodynamics [12]. The gas-liquid flows in these columns are intrinsically unstable and the dynamics of such flows influence mixing and mass transfer performance. The hydrodynamics of the liquid phase illustrates that the liquid flow rate in the aquarium is directly proportional to the injected gas flow rate. It is therefore important to characterize the dynamics of the gas-liquid flow [13]. Also, the complete knowledge of the global dynamics of the aquarium fluids is based on that of the bubble [14]. Knowledge of bubble size and vacuum ratio is crucial for determining gas-liquid interfacial air and therefore mass transfer [15]. The non-intrusive techniques used is the image analysis by PIV [11, 16]. The influence of the extinction coefficient changes was observed on CO₂ concentration during tests in the wake of a free-climbing bubble in a water column.

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